ISSN 1070-4280, Russian Journal of Organic Chemistry, 2007, Vol. 43, No. 3, pp. 426–428. © Pleiades Publishing, Ltd., 2007. Original Russian Text © A.A. Avetisyan, I.L. Aleksanyan, K.S. Sargsyan, 2007, published in Zhurnal Organicheskoi Khimii, 2007, Vol. 43, No. 3, pp. 427–429.

Synthesis of 2-Bromomethyl-4-methyl-2,3-dihydrofuro[3,2-*c*]quinoline and Some Its Transformations

A. A. Avetisyan, I. L. Aleksanyan, and K. S. Sargsyan

Erevan State University, ul. A. Manukyana 1, Erevan, 375025 Armenia e-mail: organkim@sun.ysu.am

Received January 19, 2006; revised January 4, 2007

Abstract—Electrophilic intramolecular heterocyclization of 3-(2-chloroprop-2-en-1-yl)-2-methylquinolin-4-ol by the action of bromine gave 2-bromomethyl-2-chloro-4-methyl-2,3-dihydrofuro[3,2-*c*]quinoline hydrobromide which was converted into 2-hydroxymethyl-, 2-alkoxymethyl-, and 2-dialkylaminomethyl-4-methyl-furo[3,2-*c*]quinolines by treatment with the corresponding nucleophiles.

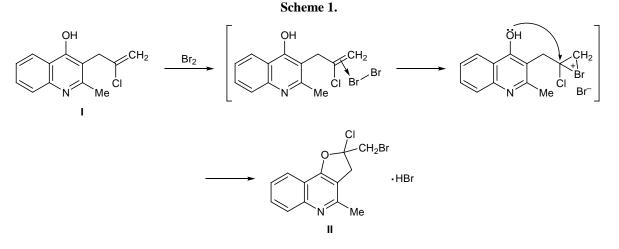
DOI: 10.1134/S1070428007030177

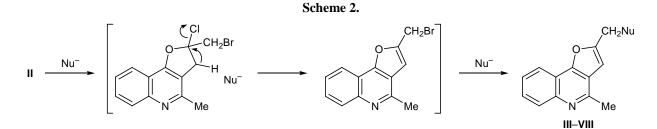
Quinoline derivatives are quite promising and accessible intermediate products for the synthesis of various polynuclear heterocyclic systems that are interesting from the viewpoint of their biological activity [1, 2]. In particular, many furo[3,2-*c*]quinoline derivatives were found to exhibit strong antiallergic, antiinflammatory, vasodilating, antiarrythmic, antispasmodic, psychotropic, bronchodilating, neurotropic, antifungal, and coronary activity [3].

While continuing our studies on transformations of 3-(2-chloroprop-2-en-1-yl)-4-hydroxyquinolines [4], we developed a convenient procedure for the synthesis of 2-bromomethyl-2-chloro-4-methyl-2,3-dihydrofuro-[3,2-*c*]quinoline hydrobromide via electrophilic intra-molecular heterocyclization of 3-(2-chloroprop-2-en-1-yl)-4-hydroxy-2-methylquinoline and examined reac-

tions of the cyclization product with a number of nucleophiles.

Initial 3-(2-chloroprop-2-en-1-yl)-4-hydroxy-2methylquinoline (**I**) was synthesized by the procedure described in [4]. Treatment of **I** with a solution of bromine in chloroform at room temperature for 1–2 h gave 2-bromomethyl-2-chloro-4-methyl-2,3-dihydrofuro[3,2-*c*]quinoline hydrobromide (**II**) (Scheme 1). The ¹H NMR spectrum of **II** contained signals at δ 3.24 (s, 2H, CH₂) and 3.62 ppm (s, 2H, CH₂Br). As shown in Scheme 1, addition of bromine at the double C=C bond of the chloroallyl group gives intermediate bromonium ion which can be stabilized in two ways: via nucleophilic addition of bromide ion or by heterocyclization involving the electron-donor hydroxy group. If the hydroxy group is sufficiently nucleo-





III, Nu = HO; IV, Nu = MeO; V, Nu = EtO; VI, Nu = i-PrO; VII, $Nu = Et_2N$; VIII, Nu = morpholino.

philic, the rate of heterocyclization exceeds the rate of nucleophilic addition.

Bromomethylfuroquinoline **II** reacted with various nucleophiles (Nu = OH, OR, R₂N) to form the corresponding 2-substituted 4-methylfuro[3,2-*c*]quinolines **III–VIII** in high yields (Scheme 2). In the first step, nucleophile promotes elimination of hydrogen chloride from the dihydrofuran ring, and next follows nucleophilic replacement of the bromine atom in the 2-bromomethyl group. Insofar as the second step is faster than the first one, intermediate 2-bromomethyl-4-methylfuro[3,2-*c*]quinoline cannot be isolated. The structure of products **III–VIII** was proved by the ¹H NMR data. The 3-H proton in **III–VIII** resonated in the ¹H NMR spectra as a singlet at δ 6.75–6.90 ppm.

2-Hydroxymethyl-, 2-alkoxymethyl-, and 2-diethylaminomethyl-4-methylfuro[3,2-c]quinolines **III–VII** were obtained by heating initial 2-bromomethyl-2chloro-4-methyl-2,3-dihydrofuro[3,2-c]quinoline hydrobromide (**II**) with 3 equiv of the corresponding nucleophile (aqueous–alcoholic alkali, sodium alkoxides, and diethylamine in dimethylformamide). In the synthesis of 2-morpholinomethyl-4-methyl-2,3-dihydrofuro[3,2-c]quinoline (**VIII**), the reaction was carried out with equimolar amounts of the reactants in DMF in the presence of anhydrous pyridine.

EXPERIMENTAL

The IR spectra were recorded on a Specord 75IR spectrometer. The ¹H NMR spectra were measured on a Varian Mercury-300 instrument (300 MHz) from solutions in DMSO- d_6 . The purity of the products was checked by TLC on Silufol UV-254 plates (development with iodine vapor).

2-Bromomethyl-2-chloro-4-methyl-2,3-dihydrofuro[3,2-c]quinoline hydrobromide (II). 3-(2-Chloroprop-2-en-1-yl)-2-methyl-4-hydroxyquinoline (I) [4], 5.84 g (0.025 mol), was dissolved in 30 ml of CHCl₃, 50 ml of a 0.5 M solution of bromine (0.025 mol) in CHCl₃ was added dropwise under vigorous stirring, and the mixture was stirred for 1 h at room temperature. The precipitate was filtered off, washed with CHCl₃, and dried. Yield 9.1 g (92%), mp 247–249°C. ¹H NMR spectrum, δ , ppm: 2.75 s (3H, CH₃), 3.24 s (2H, CH₂), 3.62 s (2H, CH₂Br), 7.5–8.2 m (4H, H_{arom}). Found, %: C 39.71; H 2.93; Cl+Br 49.81; N 3.70. C₁₃H₁₂Br₂ClNO. Calculated, %: C 39.64; H 3.04; Cl+Br 49.68; N 3.55.

(4-Methylfuro[3,2-*c*]quinolin-2-yl)methanol (III). Compound II, 0.79 g (0.002 mol), was added to 0.006 mol of an aqueous–alcoholic solution of alkali. The mixture was heated for 2 h on a water bath, the alcohol was distilled off, the residue was dissolved in a dilute acid, the solution was filtered, the filtrate was made alkaline, and the precipitate was filtered off. Yield 0.49 g (87%), mp 237–240°C, R_f 0.54 (toluene– ethanol, 2:1). ¹H NMR spectrum, δ , ppm: 2.65 s (3H, CH₃), 4.6 s (2H, CH₂), 4.90 s (1H, OH), 6.8 s (1H, 3-H), 7.5–8.2 m (4H, H_{arom}). Found, %: C 73.31; H 5.10; N 6.70. C₁₃H₁₁NO₂. Calculated, %: C 73.23; H 5.16; N 6.57.

2-Alkoxymethyl-4-methylfuro[3,2-c]quinolines IV–VI (*general procedure*). Quinoline **II**, 0.002 mol, was added to a solution of sodium alkoxide prepared from 50 ml of the corresponding anhydrous alcohol and 0.14 g (0.006 mol) of metallic sodium. The mixture was heated for 2 h on a water bath, the solvent was distilled off, the residue was dissolved in a dilute acid, the solution was filtered and neutralized to pH 8, and the precipitate was filtered off.

2-Methoxymethyl-4-methylfuro[**3**,2-*c*]**quinoline** (**IV**). Yield 0.40 g (86%), mp 74–75°C, R_f 0.62 (toluene–ethanol, 2:1). ¹H NMR spectrum, δ , ppm: 2.70 s (3H, CH₃), 3.62 s (3H, OCH₃), 4.65 s (2H, CH₂), 6.90 s (1H, 3-H), 7.5–8.2 m (4H, H_{arom}). Found, %: C 74.06; H 5.61; N 6.24. C₁₄H₁₃NO₂. Calculated, %: C 74.00; H 5.72; N 6.16.

2-Ethoxymethyl-4-methylfuro[**3,2-**c]**quinoline** (V). Yield 0.44 g (91%), mp 69–71°C, $R_{\rm f}$ 0.62

(toluene–ethanol, 2:1). ¹H NMR spectrum, δ , ppm: 1.18 t (3H, OCH₂CH₃), 2.72 s (3H, 4-CH₃), 3.40 q (2H, OCH₂CH₃), 3.85 s (2H, CH₂), 6.80 s (1H, 3-H), 7.3–8.2 m (4H, H_{arom}). Found, %: C 74.57; H 6.31; N 5.68. C₁₅H₁₅NO₂. Calculated, %: C 74.68; H 6.22; N 5.80.

2-Isopropoxymethyl-4-methylfuro[**3**,**2**-*c*]**quinoline** (**VI**). Yield 0.46 g (90%), mp 66–68°C, $R_{\rm f}$ 0.69 (toluene–ethanol, 3:1). ¹H NMR spectrum, δ , ppm: 1.20 d (6H, CH₃), 2.90 s (3H, 4-CH₃), 3.80 m (1H, OCH), 4.65 s (2H, OCH₂), 6.90 s (1H, 3-H), 7.5–8.2 m (4H, H_{arom}). Found, %: C 75.36; H 6.52; N 5.61. C₁₆H₁₇NO₂. Calculated, %: C 75.29; H 6.66; N 5.49.

N,*N*-Diethyl(4-methylfuro[3,2-*c*]quinolin-2-yl)methanamine (VII). Diethylamine, 0.65 ml (0.006 mol), was added to a solution of 0.79 g (0.002 mol) of compound **II** in 10 ml of DMF, and the mixture was heated for 3 h under reflux. The solvent was distilled off under reduced pressure, the residue was dissolved in a dilute acid, the solution was filtered, and the filtrate was neutralized to pH 8. An oily material separated and was extracted into benzene. Removal of the solvent from the extract left 0.51 g (95%) of compound **VII** as an oily substance, R_f 0.63 (toluene–ethanol, 1.8:1). ¹H NMR spectrum, δ , ppm: 1.12 m (6H, CH₃), 2.60 q (4H, CH₂), 2.82 s (3H, 4-CH₃), 4.85 s (2H, 2-CH₂), 6.8 s (1H, 3-H), 7.45–8.2 m (4H, H_{arom}). Found, %: C 76.24; H 7.37; N 10.57. C₁₇H₂₀N₂O. Calculated, %: C 76.12; H 7.46; N 10.44.

4-Methyl-2-morpholinomethylfuro[3,2-c]quinoline (VIII). Morpholine, 0.2 ml (0.002 mol), was added to a solution of 0.79 g (0.002 mol) of furoquinoline II in 10 ml of DMF containing 0.5 ml of anhydrous pyridine, and the mixture was heated for 3 h under reflux. The solvent was distilled off under reduced pressure, the residue was dissolved in a dilute acid, the solution was filtered, and the filtrate was neutralized to pH 8. An oily material separated and was extracted into diethyl ether. Removal of the solvent from the extract left 0.4 g (70%) of compound **VIII** as a crystalline substance, $R_{\rm f}$ 0.69 (toluene– ethanol, 2:1). ¹H NMR spectrum, δ, ppm: 2.52 t (4H, CH₂NCH₂), 2.80 s (3H, 4-CH₃), 3.60 t (4H, CH₂OCH₂), 3.80 s (2H, 2-CH₂), 6.83 s (1H, 3-H), 7.45-8.2 m (4H, H_{arom}). Found, %: C 72.21; H 6.43; N 10.11. C₁₇H₁₈N₂O₂. Calculated, %: C 72.34; H 6.38; N 9.93.

REFERENCES

- Orekhov, A.P., *Khimiya alkaloidov* (Chemistry of Alkaloids), Moscow: Akad. Nauk SSSR, 1955, p. 208.
- 2. Abass, M., Heterocycles, 2005, vol. 65, p. 901.
- 3. Aleksanyan, I.L., *Cand. Sci. (Chem.) Dissertation*, Erevan, 1985.
- 4. Avetisyan, A.A., Aleksanyan, I.L., and Pivazyan, A.A., *Russ. J. Org. Chem.*, 2006, vol. 42, p. 739.